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Stereoselective synthesis of α -substituted serines from protected erythrulose oximes

M. Carda,^{a,*} J. Murga,^a S. Rodríguez,^a F. González,^a E. Castillo^a and J. A. Marco^{b,*}

^a*Depart. de Q. Inorgánica y Orgánica, Univ. Jaume I, Castellón, E-12080 Castellón, Spain*

^b*Depart. de Q. Orgánica, Univ. de Valencia, E-46100 Burjassot, Valencia, Spain*

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Abstract

The additions of organolithium reagents to the C=N bond of several chiral oxime ethers derived from erythrulose afforded protected amino polyols with high diastereoselectivity. Four of the latter compounds have been converted into the α,α -disubstituted α -amino acids (*R*)-2-(–)-methylserine, (*S*)-2-(+)-methylserine, (*R*)-(+)-2-phenylserine and (*R*)-(–)-2-*n*-butylserine. © 1998 Elsevier Science Ltd. All rights reserved.

1. Introduction

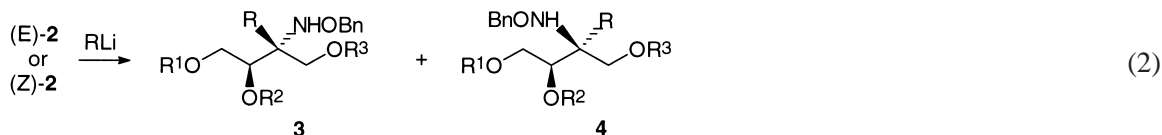
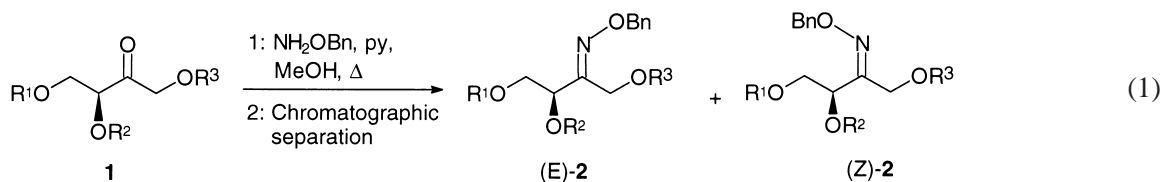
Amino polyols¹ and non-proteinaceous amino acids² are two types of nitrogen-containing compounds of particular biological relevance. The latter are secondary metabolites occurring in many living organisms³ and also constitute part of the structure of many peptidic antibiotics.⁴ Furthermore, they are used for the synthesis of modified peptides, which display useful properties as enzyme inhibitors and peptidomimetics.⁵ Among the non-proteinaceous amino acids, α,α -disubstituted α -amino carboxylic acids, which contain a nitrogen bound to a quaternary carbon atom, have attracted particular interest.⁶ Peptides which incorporate this type of amino acid are characterized by a higher conformational rigidity and by an enhanced stability towards hydrolysis.^{6–8} There are not many methods of preparing such compounds in enantiopure form. Most of them rely on the alkylation of chiral amino acid enolate equivalents.⁹ This fact places some limits on their applicability, as certain α -amino acids, such as those with α -*tert*-alkyl or α -aryl substituents, cannot be easily prepared in this way. Other methods include aldol reactions with amino ester enolates,¹⁰ stereoselective rearrangements,¹¹ ring-opening of suitably substituted β -lactams,^{7g,12} enzymatic resolutions,¹³ etc. Very few procedures, aside from the Strecker reaction and its variants,¹⁴ rely on stereoselective additions of carbon nucleophiles to C=N bonds.¹⁵ We have shown that the additions of organolithium reagents to chiral ketoxime ethers (*E*)/(*Z*)-2

* Corresponding authors. E-mail: alberto.marco@uv.es

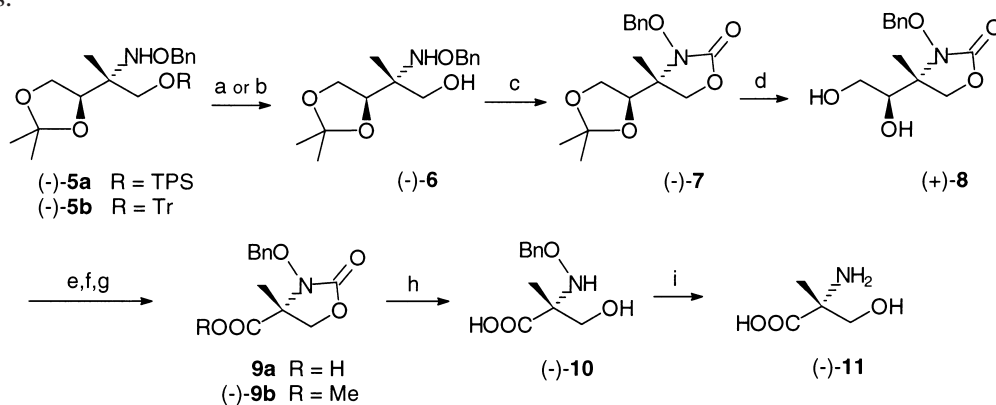
(R^1 – R^3 =protecting groups) prepared from L-erythrose derivatives of general formula **1** (Eq. 1), often take place with high stereocontrol to yield the diastereoisomeric, differentially protected amino polyols **3** and/or **4** (Eq. 2).¹⁶ These amino alcohols can then be converted through standard synthetic manipulations into enantiopure amino polyols and α,α -disubstituted α -amino acids, including those not easily available by previous methodologies. In the present paper, we describe in full the preparation of the α,α -disubstituted α -amino acids (*R*)-2-(–)-methylserine, (*S*)-2-(+)-methylserine, (*R*)-(+)-2-phenylserine and (*R*)-(–)-2-*n*-butylserine by means of this methodology.

2. Results and discussion

The starting materials were protected amino polyols **3** (R =Me, *n*Bu, Ph) obtained with a high diastereoisomeric ratio (d.r. >95:5) by addition of the corresponding organolithium reagents to oximes (*E*)-**2**.¹⁶



Various protecting groups R^1 – R^3 have been used for this purpose. Amino polyol acetonides **3** (R^1 , R^2 =acetonide and R^3 =*tert*-butyldiphenylsilyl, TPS, or trityl, Tr) with R =Me, *n*Bu and Ph, obtained with a high d.r. from the appropriate oximes (*E*)-**2**,¹⁶ were selected for transformation into α -substituted serines.¹⁷ In the case of (*R*)-2-(–)-methylserine, for instance, compounds (–)-**5a**/(–)-**5b** were the starting materials.

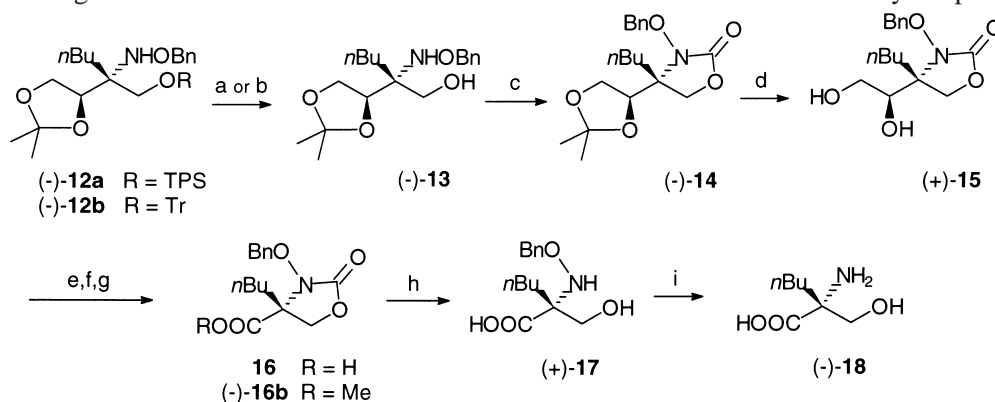


Reaction conditions. a) TBAF, THF, RT. b) TFA/TFAA, CH_2Cl_2 , RT. c) CDI, C_6H_6 , Δ . d) PPTS, aq MeOH, Δ . e) NaIO_3 , aq THF, RT. f) NaClO_2 , NaH_2PO_4 , aq *t*BuOH. g) CH_2N_2 . h) 2M NaOH, aq EtOH. i) H_2 , Pd/C, MeOH.

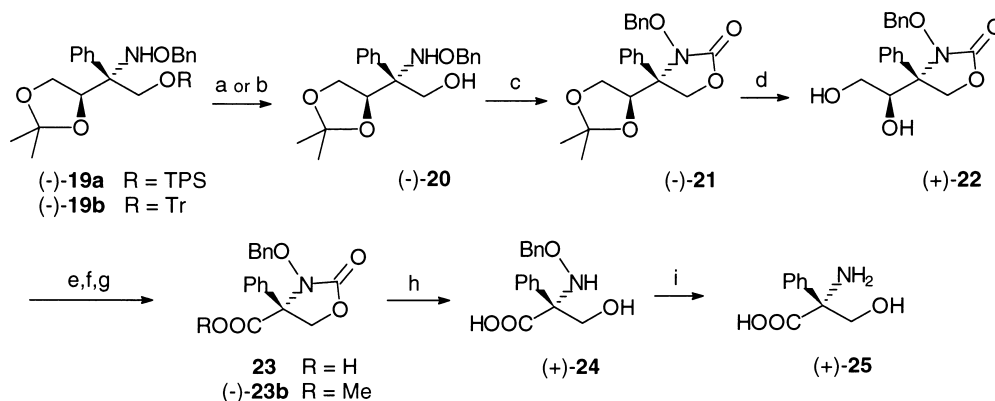
Desilylation of (–)-**5a**, as well as detritylation of (–)-**5b**, provided the protected amino polyol (–)-**6**,^{16b} which still contained a small amount of the minor stereoisomer. Treatment of the latter with carbonyldiimidazole (CDI) afforded oxazolidinone (–)-**7** and its minor stereoisomer, which could now be separated by chromatography on silica gel. Acetonide cleavage catalyzed by pyridinium *p*-

toluenesulphonate (PPTS) gave diol (+)-**8**, which was then oxidized via a two-step procedure¹⁸ to acid **9a**, isolated and purified as the methyl ester (–)-**9b**. Basic hydrolysis removed both the oxazolidinone and methyl ester groups and furnished N-benzyloxy amino acid (–)-**10**. Reductive cleavage of the N–O bond was best achieved via hydrogenolysis with a Pd/C catalyst to yield (*R*)-2-methylserine (–)-**11**. Physical and spectral data of the obtained product (see Experimental) were essentially coincident with those reported in the literature.^{9a,b,12a,14a}

The amino acids (*R*)-2-*n*-butylserine (–)-**18** and (*R*)-2-phenylserine (+)-**25** were prepared along the same lines as for the synthesis of (–)-**11**. The only difference was the means of cleavage of the acetonide moiety, which was performed by transacetalization with ethanedithiol and *p*-toluenesulphonic acid.¹⁹ The preparation of (*R*)-2-*n*-butylserine has not been previously described in the literature. The amino acid 2-phenylserine has previously been described only in racemic form,²⁰ although a derivative with unknown configuration was obtained with 67% enantiomeric excess via a metal-catalyzed process.²¹



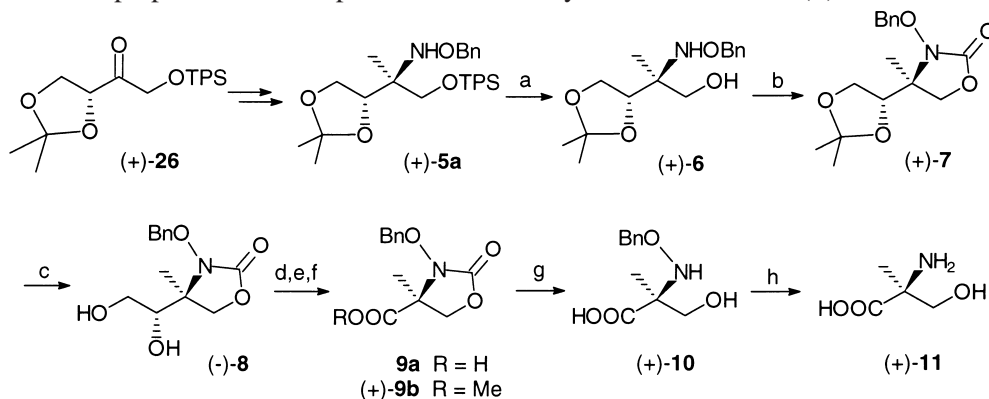
Reaction conditions. a) TBAF, THF, RT. b) TFA/TFAA, CH₂Cl₂, RT. c) CDI, C₆H₆, Δ (85% overall). d) (CH₂SH)₂, TsOH, CHCl₃, Δ. e) NaIO₄, aq THF. f) NaClO₂, aq *t*BuOH. g) CH₂N₂ (40% overall). h) aq NaOH/EtOH, RT. i) H₂, Pd/C, MeOH (50% overall).



Reaction conditions. a) TBAF, THF, RT. b) TFA/TFAA, CH₂Cl₂, RT. c) CDI, C₆H₆, Δ (85% overall). d) (CH₂SH)₂, TsOH, CHCl₃, Δ. e) NaIO₄, aq THF. f) NaClO₂, aq *t*BuOH. g) CH₂N₂ (40% overall). h) aq NaOH/EtOH, RT. i) H₂, Pd/C, MeOH (50% overall).

The methodology described in this paper does not only allow the synthesis of serine derivatives with the (*R*)-configuration. Their antipodes with the opposite (*S*)-configuration may also be obtained by using D-erythrose precursors as the starting materials.²² Therefore, we also synthesized (*S*)-2-methylserine

(+)-**11** from compound (+)-**5a** by the same series of reactions described above for (–)-**11**. Compound (+)-**5a** was in turn prepared in two steps^{16b} from the D-erythrose derivative (+)-**26**.²²



Reaction conditions. a) TBAF, THF, RT. b) CDI, C₆H₆, Δ (85% overall). c) PPTS, aq MeOH, Δ. d) NaIO₄, aq THF. e) NaClO₂, aq *t*BuOH. f) CH₂N₂ (40% overall). g) aq NaOH/EtOH, RT. h) H₂, Pd/C, MeOH (50% overall).

The methodology we have described above is therefore very useful for the preparation of many types of α,α -disubstituted α -amino acids in enantiopure form, including those not easily available by alternative procedures. Furthermore, the protected amino polyols obtained with this procedure possess several hydroxyl groups, a fact which opens the way to the synthesis of other biologically relevant, polyfunctionalized compounds, such as polyhydroxylated amino acids,²³ diamino acids,²⁴ N-hydroxy amino acids,²⁵ and branched aminosugars.²⁶ Finally, since D-erythrose derivatives are also easily available,²² all these compounds can be prepared in either antipodal form. Further research in these directions is in progress.

3. Experimental

For general experimental details, see the preceding article.^{16b} All obtained products gave satisfactory microanalytical data (C, H, $\pm 0.5\%$).

3.1. (4R)-3-Benzyloxy-4-methyl-4-[(4R)-2,2-dimethyl-1,3-dioxolan-4-yl]-1,3-oxazolidin-2-one, (–)-7

Amino alcohol (–)-**6**^{16b} (1.406 g, 5 mmol) was dissolved in dry benzene (40 ml) and treated with CDI (973 mg, 6 mmol). The reaction mixture was heated at reflux under Ar for 4 h. Work-up (Et₂O) and column chromatography (hexane:EtOAc=4:1) afforded (–)-**7** (1.25 g, 81%) together with its minor diastereomer (100 mg, ca. 6%). Oxazolidinone (–)-**7** was isolated as a white solid, mp 99–100°C, $[\alpha]_D -7.2$ (CHCl₃, *c* 1); IR ν_{\max} cm⁻¹: 1782 (C=O); EIMS, *m/z* 307.1411 (M⁺). Calcd for C₁₆H₂₁NO₅, M=307.1420; ¹H NMR: δ 7.45–7.30 (5H, *m*, arom.), 5.01, 4.95 (2H, AB system, J=11 Hz, NOCH₂Ph), 4.31 (1H, *d*, J=8.5 Hz, H-5_a), 4.21 (1H, *dd*, J=7.2, 5.5 Hz, H-4'), 3.96 (1H, *dd*, J=9, 7.2 Hz, H-5'_a), 3.86 (1H, *d*, J=8.5 Hz, H-5_b), 3.67 (1H, *dd*, J=9, 5.5 Hz, H-5'_b), 1.41, 1.29 (2×3H, 2×*s*, acetonide Me), 0.98 (3H, *s*, MeC₄); ¹³C NMR: δ 158.5 (C-2), 135.3 (arom. C), 129.9, 128.9, 128.5 (arom. CH), 109.8 (acetonide C), 78.6 (NOCH₂Ph), 76.0 (C-4'), 68.8 (C-5), 68.4 (C-5'), 63.4 (C-4), 25.9, 24.3 (2×acetonide Me), 17.1 (MeC₄). Primed numbers in NMR data correspond to atoms of the dioxolane ring. Anal. calcd for C₁₆H₂₁NO₅: C, 62.53; H, 6.89; N, 4.56. Found: C, 62.63; H, 7.00; N, 4.31.

3.2. (4S)-3-Benzoyloxy-4-methyl-4-[(4S)-2,2-dimethyl-1,3-dioxolan-4-yl]-1,3-oxazolidin-2-one, (+)-7

Obtained from amino alcohol (+)-6^{16b} under the same conditions as for (–)-7: $[\alpha]_D +7$ (CHCl₃, *c* 1.8). The other physical properties are identical to those of (–)-7. Anal. calcd for C₁₆H₂₁NO₅: C, 62.53; H, 6.89; N, 4.56. Found: C, 62.75; H, 6.96; N, 4.44.

3.3. (4R)-3-Benzoyloxy-4-methyl-4-[(1R)-1,2-dihydroxyethyl]-1,3-oxazolidin-2-one, (+)-8

Oxazolidinone (–)-7 (1.23 g, 4 mmol) and PPTS (50 mg, 0.2 mmol) were dissolved in a 9:1 MeOH:H₂O mixture (20 ml) and heated at reflux for 12 h. Work-up^{16b} and column chromatography (hexane:EtOAc=1:1) afforded (+)-8 (987 mg, 92%) as a white solid, mp 118–120°C, $[\alpha]_D +17.4$ (MeOH, *c* 2.7); IR ν_{\max} cm⁻¹: 3400 (br, OH), 1764 (C=O); FABMS, *m/z* 268.1183 (M+H⁺). Calcd for C₁₃H₁₈NO₅, M=268.1185; ¹H NMR: δ 7.45–7.30 (5H, *m*, arom.), 5.04, 4.95 (2H, AB system, J=11 Hz, NOCH₂Ph), 4.43 (1H, *d*, J=8.8 Hz, H-5), 3.86 (1H, *d*, J=8.8 Hz, H-5'), 3.65–3.40 (3H, *m*, CH₂OHCHOH), 2.50 (2H, *br s*, 2 OH), 1.15 (3H, *s*, MeC₄); ¹³C NMR: δ 159.4 (C-2), 135.3 (arom. C), 130.0, 129.3, 128.8 (arom. CH), 77.9 (NOCH₂Ph), 71.6 (CHOH), 68.5 (C-5), 65.7 (C-4), 62.1 (CH₂OH), 18.2 (MeC₄). Anal. calcd for C₁₃H₁₇NO₅: C, 58.42; H, 6.41; N, 5.24. Found: C, 58.63; H, 6.60; N, 5.31.

3.4. (4S)-3-Benzoyloxy-4-methyl-4-[(1S)-1,2-dihydroxyethyl]-1,3-oxazolidin-2-one, (–)-8

Obtained from oxazolidinone (+)-7 under the same conditions as for (+)-8: $[\alpha]_D -16.8$ (MeOH, *c* 2.2). The other physical properties are identical to those of (+)-8. Anal. calcd for C₁₃H₁₇NO₅: C, 58.42; H, 6.41; N, 5.24. Found: C, 58.55; H, 6.30; N, 5.11.

3.5. (4R)-3-Benzoyloxy-4-methyl-4-methoxycarbonyl-1,3-oxazolidin-2-one, (–)-9b

Diol (+)-8 (962 mg, 3.6 mmol) was dissolved in THF (10 ml) and treated with a solution of NaIO₄ (1.54 g, 7.2 mmol) in water (2 ml). The reaction mixture was stirred at room temperature for 2 h. Work-up (EtOAc) afforded an oily material which was dissolved in *t*BuOH (70 ml) and treated successively with 2-methyl-2-butene (20 ml), 25% aqueous NaClO₂ (13 ml, ca. 36 mmol) and NaH₂PO₄ (3.8 g). The reaction mixture was stirred at room temperature for 12 h. Work-up (EtOAc) afforded an oily material which was directly esterified with ethereal diazomethane. Solvent removal *in vacuo* and column chromatography (hexane:EtOAc=4:1) provided (–)-9b (716 mg, 75% overall) as an oil, $[\alpha]_D -9.4$ (CHCl₃, *c* 1.7); IR ν_{\max} cm⁻¹: 1793, 1747 (C=O); EIMS, *m/z* 266.1027 (M+H⁺, 1), 206 (M⁺–COOMe, 30), 100 (45), 91 (100). Calcd for C₁₃H₁₆NO₅, M=266.1028; ¹H NMR: δ 7.45–7.30 (5H, *m*, arom.), 5.06, 4.99 (2H, AB system, J=10.7 Hz, NOCH₂Ph), 4.31 (1H, *d*, J=9 Hz, H-5), 4.00 (1H, *d*, J=9 Hz, H-5'), 3.77 (3H, *s*, OMe), 1.36 (3H, *s*, MeC₄); ¹³C NMR: δ 170.4 (ester C=O), 157.9 (C-2), 135.1 (arom. C), 129.8, 128.9, 128.4 (arom. CH), 78.8 (NOCH₂Ph), 69.5 (C-5), 65.4 (C-4), 53.2 (OMe), 18.3 (MeC₄). Anal. calcd for C₁₃H₁₅NO₅: C, 58.86; H, 5.70; N, 5.28. Found: C, 58.59; H, 5.55; N, 5.15.

3.6. (4S)-3-Benzoyloxy-4-methyl-4-methoxycarbonyl-1,3-oxazolidin-2-one, (+)-9b

Obtained from diol (–)-8 under the same conditions as for (–)-9b: $[\alpha]_D +9.0$ (CHCl₃, *c* 1.8). The other physical properties are identical to those of (–)-9b. Anal. calcd for C₁₃H₁₅NO₅: C, 58.86; H, 5.70; N, 5.28. Found: C, 58.67; H, 5.81; N, 5.35.

3.7. (2R)-3-Hydroxy-2-(N-benzyloxyamino)-2-methylpropionic acid, (–)-**10**

Methyl ester (–)-**9b** (663 mg, 2.5 mmol) was dissolved in EtOH (10 ml) and treated with a 2 M aqueous solution of sodium hydroxide (10 ml). The reaction mixture was stirred at room temperature for 12 h. After this time, most of the EtOH was removed *in vacuo* and the pH of the solution was adjusted to 1 with 5 M aq HCl. Work-up (EtOAc) and column chromatography (EtOAc:MeOH=1:1) provided (–)-**10** (366 mg, 65%) as an oil, $[\alpha]_D -5.4$ (MeOH, *c* 0.2); IR ν_{\max} cm^{-1} : 3300 (br, OH), 1591, 1414 (COO[–]); CIMS, *m/z* 194.0815 (M⁺–CH₂OH). Calcd for C₁₀H₁₂NO₃, M=194.0817; ¹H NMR (CD₃OD): δ 7.40–7.20 (5H, *m*, arom.), 4.70 (2H, *s*, NOCH₂Ph), 3.75 (1H, *d*, J=10.7 Hz, H-3), 3.63 (1H, *d*, J=10.7 Hz, H-3'), 1.20 (3H, *s*, MeC₂); ¹³C NMR (CD₃OD): δ 181.0 (C-1), 139.3 (arom. C), 129.3, 129.2, 128.6 (arom. CH), 77.5 (NOCH₂Ph), 67.4 (C-2), 65.9 (C-3), 19.0 (MeC₂). Anal. calcd for C₁₁H₁₅NO₄: C, 58.66; H, 6.71; N, 6.22. Found: C, 58.38; H, 6.56; N, 6.05.

3.8. (2S)-3-Hydroxy-2-(N-benzyloxyamino)-2-methylpropionic acid, (+)-**10**

Obtained from methyl ester (+)-**9b** under the same conditions as for (–)-**10**: $[\alpha]_D +5.0$ (MeOH, *c* 0.3). The other physical properties are identical to those of (–)-**10**. Anal. calcd for C₁₁H₁₅NO₄: C, 58.66; H, 6.71; N, 6.22. Found: C, 58.87; H, 6.76; N, 6.30.

3.9. (R)-(–)-2-Methyl serine, (–)-**11**

Pd/C catalyst (50 mg) was suspended in MeOH (3 ml) and stirred under an H₂ atmosphere for 10 min. Acid (–)-**10** (225 mg, 1 mmol) dissolved in MeOH (15 ml) was then injected via syringe. The mixture was stirred under the H₂ atmosphere at room temperature for 48 h. After this time, the catalyst was removed by filtration and the solvent was removed *in vacuo*. Aqueous 1 M HCl was added to the residue until pH 1. The solution was then evaporated to dryness and the residue was purified through a Dowex 50WX8-400 column according to the described procedure.^{9b} This yielded (R)-(–)-2-methyl serine, (–)-**11** (65 mg, 55%) as a white solid, mp 225–230°C (dec.), lit.^{9a} mp 240–245°C (dec.), lit.^{12a} mp 235–245°C (dec.); $[\alpha]_D -6.3$ (H₂O, *c* 1.1), lit.^{9a} $[\alpha]_D -5.8$ (H₂O, *c* 0.289), lit.^{12a} $[\alpha]_D -6.1$ (H₂O, *c* 0.9); IR ν_{\max} cm^{-1} : 3400 (br, OH), 1650, 1450, 1410, 1380, 1275, 1088, 1050, 880; CIMS, *m/z* 120.0692 (M+H⁺). Calcd for C₄H₁₀NO₃, M=120.0660; ¹H NMR (D₂O): δ 3.95 (1H, *d*, J=12 Hz, H-3), 3.70 (1H, *d*, J=12 Hz, H-3'), 1.45 (3H, *s*, Me-C₂); ¹³C NMR (D₂O): δ 177.2 (C-1), 66.3 (C-3), 63.1 (C-2), 19.9 (Me-C₂).

3.10. (S)-(+)-2-Methyl serine, (+)-**11**

Obtained from acid (+)-**10** under the same conditions as for (–)-**11**: $[\alpha]_D +6.1$ (H₂O, *c* 0.9). The other physical properties are identical to those of (–)-**11**.

3.11. (4R)-3-Benzyloxy-4-n-butyl-4-[(4R)-2,2-dimethyl-1,3-dioxolan-4-yl]-1,3-oxazolidin-2-one, (–)-**14**

Obtained in 75% yield from amino alcohol (–)-**13**^{16b} under the same conditions as for (–)-**7**: white solid, mp 85–87°C, $[\alpha]_D -8.5$ (CHCl₃, *c* 1.2); IR ν_{\max} cm^{-1} : 1782 (C=O); CIMS, *m/z* 350.1971 (M+H⁺). Calcd for C₁₉H₂₈NO₅, M=350.1967; ¹H NMR: δ 7.40–7.30 (5H, *m*, arom.), 5.00 (2H, *s*, NOCH₂Ph), 4.27 (1H, *d*, J=8.8 Hz, H-5_a), 4.24 (1H, *dd*, J=7, 6 Hz, H-4'), 4.03 (1H, *d*, J=8.8 Hz,

H-5_b), 3.95 (1H, *dd*, J=9, 7 Hz, H-5'_a), 3.62 (1H, *dd*, J=9, 6 Hz, H-5'_b), 1.60 and 1.30–1.15 (6H, *m*, CH₂CH₂CH₂), 1.43, 1.30 (2×3H, 2×*s*, acetonide Me), 0.85 (3H, *t*, J=7 Hz, MeCH₂); ¹³C NMR: δ 158.4 (C-2), 135.3 (arom. C), 129.3, 128.7, 128.4 (arom. CH), 109.7 (acetonide C), 78.4 (NOCH₂Ph), 76.4 (C-4'), 65.9, 64.7 (C-5, C-5'), 65.7 (C-4), 30.8 (CH₂C₃H₇), 25.9, 24.4 (2×acetonide Me), 24.9, 22.9 (CH₂CH₂), 13.8 (MeCH₂). Primed numbers in NMR data correspond to atoms of the dioxolane ring. Anal. calcd for C₁₉H₂₇NO₅: C, 65.31; H, 7.79; N, 4.01. Found: C, 65.40; H, 7.66; N, 4.14.

3.12. (4R)-3-Benzoyloxy-4-n-butyl-4-[(1R)-1,2-dihydroxyethyl]-1,3-oxazolidin-2-one, (+)-**15**

Oxazolidinone (–)-**14** (1.048 g, 3 mmol), 1,2-ethane dithiol (2.5 ml, ca. 30 mmol) and *p*-toluenesulphonic acid monohydrate (19 mg, 0.1 mmol) were dissolved in dry CHCl₃ (25 ml) and heated at reflux for 2 h. Work-up (CH₂Cl₂) and column chromatography (hexane:EtOAc=1:1) furnished (+)-**15** (753 mg, 81%) as a white solid, mp 74–76°C, [α]_D +54.4 (CHCl₃, *c* 1.7); IR ν_{max} cm⁻¹: 3400 (br, OH), 1765 (C=O); FABMS, *m/z* 310.1642 (M+H⁺). Calcd for C₁₆H₂₄NO₅, M=310.1655; ¹H NMR: δ 7.45–7.35 (5H, *m*, arom.), 5.10, 5.00 (2H, AB system, J=11.3 Hz, NOCH₂Ph), 4.41 (1H, *d*, J=9 Hz, H-5), 3.96 (1H, *d*, J=9 Hz, H-5'), 3.60 (1H, *m*, CH₂OH), 3.40 (2H, *m*, CH₂OH, CH₂OH), 2.30, 2.10 (2H, 2×*br s*, 2×OH), 1.60–1.20 (6H, *m*, CH₂CH₂CH₂), 0.89 (3H, *t*, J=7.2 Hz, Me); ¹³C NMR: δ 159.6 (C-2), 135.4 (arom. C), 129.8, 129.4, 128.9 (arom. CH), 77.3 (NOCH₂Ph), 71.8 (CHOH), 67.9 (C-4), 66.7 (C-5), 62.0 (CH₂OH), 32.5 (CH₂C₃H₇), 24.9, 23.0 (CH₂CH₂), 13.9 (Me). Anal. calcd for C₁₆H₂₃NO₅: C, 62.12; H, 7.49; N, 4.53. Found: C, 62.33; H, 7.46; N, 4.34.

3.13. (4R)-3-Benzoyloxy-4-n-butyl-4-methoxycarbonyl-1,3-oxazolidin-2-one, (–)-**16b**

Obtained in 90% yield from diol (+)-**15** under the same conditions as for (–)-**9b**: oil, [α]_D –6.5 (CHCl₃, *c* 1.3); IR ν_{max} cm⁻¹: 1790, 1746 (C=O); EIMS, *m/z* 308.1494 (M+H⁺, 1), 248 (M⁺–COOMe, 8), 142 (15), 91 (100). Calcd for C₁₆H₂₂NO₅, M=308.1498; ¹H NMR: δ 7.45–7.35 (5H, *m*, arom.), 5.10, 5.06 (2H, AB system, J=10.3 Hz, NOCH₂Ph), 4.38 (1H, *d*, J=9 Hz, H-5), 4.16 (1H, *d*, J=9 Hz, H-5'), 3.78 (3H, *s*, OMe), 1.85 (2H, *m*, CH₂C₃H₇), 1.30–1.20 (4H, *m*, CH₂CH₂), 0.86 (3H, *t*, J=7 Hz, MeCH₂); ¹³C NMR: δ 170.4 (ester C=O), 158.0 (C-2), 135.1 (arom. C), 129.5, 128.7, 128.4 (arom. CH), 78.7 (NOCH₂Ph), 67.9 (C-4), 67.3 (C-5), 53.0 (OMe), 32.1 (CH₂C₃H₇), 24.9, 22.7 (CH₂CH₂), 13.7 (Me). Anal. calcd for C₁₆H₂₁NO₅: C, 62.53; H, 6.89; N, 4.56. Found: C, 62.43; H, 7.01; N, 4.54.

3.14. (2R)-2-(*N*-Benzoyloxyamino)-2-(hydroxymethyl)hexanoic acid, (+)-**17**

Obtained in 73% yield from methyl ester (–)-**16b** under the same conditions as for (–)-**10**: white solid, mp 120–125°C (dec.), [α]_D +1.7 (CHCl₃, *c* 0.3); IR ν_{max} cm⁻¹: 3300 (br, OH), 1586, 1426, 1264, 1091, 874; FABMS, *m/z* 268.1538 (M+H⁺). Calcd for C₁₄H₂₂NO₄, M=268.1549; ¹H NMR (CD₃OD): δ 7.40–7.20 (5H, *m*, arom.), 4.75, 4.71 (2H, AB system, J=11.5 Hz, NOCH₂Ph), 3.79 (1H, *d*, J=10.7 Hz, CH₂OH), 3.67 (1H, *d*, J=10.7 Hz, CH₂OH), 1.55 (2H, *m*, H-3), 1.40–1.20 (4H, *m*, H-4, H-5), 0.87 (3H, *t*, J=7 Hz, H-6); ¹³C NMR (CD₃OD): δ 180.5 (C-1), 139.0 (arom. C), 129.4, 129.3, 128.8 (arom. CH), 77.5 (NOCH₂Ph), 70.9 (C-2), 64.2 (CH₂OH), 31.6 (C-3), 26.9, 24.3 (C-4, C-5), 14.5 (C-6). Anal. calcd for C₁₄H₂₁NO₄: C, 62.90; H, 7.92; N, 5.24. Found: C, 62.83; H, 7.61; N, 5.46.

3.15. (R)-(-)-2-n-Butyl serine, (-)-18

Obtained in 50% yield from acid (+)-17 under the same conditions as for (-)-11: white solid, mp 225–232°C (dec.); $[\alpha]_D -11.8$ (H₂O, *c* 0.3); IR ν_{\max} cm⁻¹: 3300 (br, OH), 1625, 1449, 1092; FABMS, *m/z* 162.1133 (M+H⁺). Calcd for C₇H₁₆NO₃, M=162.1130; ¹H NMR (D₂O:CD₃OD=4:1): δ 3.90 (1H, *d*, J=11.5 Hz, CH₂OH), 3.66 (1H, *d*, J=11.5 Hz, CH₂OH), 1.80–1.60 (2H, *m*, CH₂C₃H₇), 1.50–1.15 (4H, *m*, CH₂CH₂), 0.87 (3H, *t*, J=7 Hz, Me); ¹³C NMR (D₂O:CD₃OD=4:1): δ 175.6 (C-1), 67.3 (C-2), 65.4 (CH₂OH), 32.9 (C-3), 25.9, 23.0 (C-4, C-5), 13.9 (C-6). Anal. calcd for C₇H₁₅NO₃: C, 52.16; H, 9.38; N, 8.69. Found: C, 52.33; H, 9.60; N, 8.56.

3.16. (4R)-3-Benzoyloxy-4-phenyl-4-[(4R)-2,2-dimethyl-1,3-dioxolan-4-yl]-1,3-oxazolidin-2-one, (-)-21

Obtained in 93% yield from amino alcohol (-)-20^{16b} under the same conditions as for (-)-7: white solid, mp 90–92°C, $[\alpha]_D -36.2$ (CHCl₃, *c* 1.4); IR ν_{\max} cm⁻¹: 1788 (C=O); FABMS, *m/z* 370.1651 (M+H⁺). Calcd for C₂₁H₂₄NO₅, M=370.1655; ¹H NMR: δ 7.40–7.20 (10H, *m*, arom.), 5.07, 4.72 (2H, AB system, J=10.2 Hz, NOCH₂Ph), 4.94 (1H, *dd*, J=7.5, 4.8 Hz, H-4'), 4.67 (1H, *d*, J=9 Hz, H-5_a), 4.44 (1H, *d*, J=9 Hz, H-5_b), 4.12 (1H, *dd*, J=9.3, 7.4 Hz, H-5'_a), 3.65 (1H, *dd*, J=9.3, 4.8 Hz, H-5'_b), 1.49, 1.35 (2×3H, 2×*s*, acetonide Me); ¹³C NMR: δ 158.3 (C-2), 136.8, 134.8 (arom. C), 129.3, 128.9, 128.6, 128.3, 126.9 (arom. CH), 110.4 (acetonide C), 78.6 (NOCH₂Ph), 74.8 (C-4'), 68.5 (C-5), 68.3 (C-4), 65.4 (C-5'), 25.7, 24.0 (2×acetonide Me). Primed numbers in NMR data correspond to atoms of the dioxolane ring. Anal. calcd for C₂₁H₂₃NO₅: C, 68.28; H, 6.28; N, 3.79. Found: C, 68.39; H, 6.44; N, 3.56.

3.17. (4R)-3-Benzoyloxy-4-phenyl-4-[(1R)-1,2-dihydroxyethyl]-1,3-oxazolidin-2-one, (+)-22

Obtained in 66% yield from oxazolidinone (-)-21 under the same conditions as for (+)-15: white solid, mp 68–70°C, $[\alpha]_D +24.0$ (CHCl₃, *c* 0.7); IR ν_{\max} cm⁻¹: 3400 (br, OH), 1778 (C=O); CIMS, *m/z* 330.1345 (M+H⁺). Calcd for C₁₈H₂₀NO₅, M=330.1341; ¹H NMR: δ 7.50–7.20 (10H, *m*, arom.), 5.15, 4.87 (2H, AB system, J=10.2 Hz, NOCH₂Ph), 4.44 (1H, *t*, J=4 Hz, CHOH), 4.38 (1H, *br d*, J=12 Hz, H-5_a), 4.26 (1H, *br d*, J=12 Hz, H-5_b), 4.00 (2H, *br d*, J=4 Hz, CH₂OH), 3.30 (2H, *br s*, 2 OH); ¹³C NMR: δ 159.6 (C-2), 137.2, 134.8 (arom. C), 129.2, 129.0, 128.7, 128.6, 128.3, 126.1 (arom. CH), 82.7 (CHOH), 78.4 (NOCH₂Ph), 71.3 (C-4), 60.6 (C-5), 58.6 (CH₂OH). Anal. calcd for C₁₈H₁₉NO₅: C, 65.64; H, 5.81; N, 4.25. Found: C, 65.42; H, 5.70; N, 4.44.

3.18. (4R)-3-Benzoyloxy-4-phenyl-4-methoxycarbonyl-1,3-oxazolidin-2-one, (-)-23b

Obtained in 61% overall yield from diol (+)-22 under the same conditions as for (-)-9b: oil, $[\alpha]_D -17.7$ (CHCl₃, *c* 1.3); IR ν_{\max} cm⁻¹: 1794, 1753 (C=O); EIMS, *m/z* 328.1175 (M+H⁺, 1), 312 (M⁺-Me, 1), 268 (M⁺-COOMe, 38), 162 (20), 91 (100). Calcd for C₁₈H₁₈NO₅, M=328.1185; ¹H NMR: δ 7.40–7.30 (10H, *m*, arom.), 5.19, 5.05 (2H, AB system, J=10 Hz, NOCH₂Ph), 4.71 (1H, *d*, J=9 Hz, H-5), 4.54 (1H, *d*, J=9 Hz, H-5'), 3.86 (3H, *s*, OMe); ¹³C NMR: δ 169.5 (ester C=O), 157.1 (C-2), 134.8, 134.3 (arom. C), 129.3, 129.0, 128.6, 128.3, 126.5 (arom. CH), 78.9 (NOCH₂Ph), 71.1 (C-4), 69.9 (C-5), 53.3 (OMe). Anal. calcd for C₁₈H₁₇NO₅: C, 66.05; H, 5.23; N, 4.28. Found: C, 66.22; H, 5.40; N, 4.49.

3.19. (2R)-3-Hydroxy-2-(N-benzyloxyamino)-2-phenylpropionic acid, (+)-24

Obtained in 70% yield from methyl ester (–)-23b under the same conditions as for (–)-10: white solid, mp 144–146°C, $[\alpha]_D -12.5$ (MeOH, *c* 1.7); IR ν_{\max} cm^{-1} : 3300 (br, OH), 1589, 1412, 1091, 913, 874; FABMS, *m/z* 288.1227 (M+H⁺). Calcd for C₁₆H₁₈NO₄, M=288.1236; ¹H NMR (CD₃OD): δ 7.45 (2H, *m*, arom.), 7.30–7.15 (8H, *m*, arom.), 4.61, 4.53 (2H, AB system, J=10.5 Hz, NOCH₂Ph), 4.25 (1H, *d*, J=10.5 Hz, H-3), 4.07 (1H, *d*, J=10.5 Hz, H-3); ¹³C NMR (CD₃OD): δ 180.5 (C-1), 140.6, 138.8 (arom. C), 129.6, 129.5, 129.3, 129.0, 128.8, 128.7, 128.4 (arom. CH), 77.7 (NOCH₂Ph), 73.8 (C-2), 65.1 (C-3). Anal. calcd for C₁₆H₁₇NO₄: C, 66.89; H, 5.96; N, 4.88. Found: C, 66.72; H, 5.70; N, 4.69.

3.20. (R)-(+)-2-Phenyl serine, (+)-25

Obtained in 60% yield from acid (+)-24 under the same conditions as for (–)-11: white solid, mp 248–250°C (dec.); $[\alpha]_D +19.5$ (H₂O, *c* 0.3); IR ν_{\max} cm^{-1} : 3400 (br, OH), 1624, 1407, 1087, 875; FABMS, *m/z* 182.0822 (M+H⁺). Calcd for C₉H₁₂NO₃, M=182.0817; ¹H NMR (D₂O:CD₃OD=4:1): δ 7.55–7.40 (5H, *m*, arom.), 4.35 (1H, *d*, J=12 Hz, H-3), 4.24 (1H, *d*, J=12 Hz, H-3); ¹³C NMR (D₂O:CD₃OD=4:1): δ 174.3 (C-1), 135.6 (arom. C), 130.3, 130.2, 126.8 (arom. CH), 68.5 (C-2), 64.5 (C-3). Anal. calcd for C₉H₁₁NO₃: C, 59.66; H, 6.12; N, 7.73. Found: C, 59.45; H, 6.38; N, 7.58.

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References

1. (a) Jäger, V.; Hümmer, W.; Stahl, U.; Gracza, T. *Synthesis* **1991**, 769–776, and references cited therein. (b) Ohfuné, Y. *Acc. Chem. Res.* **1992**, 25, 360–366.
2. (a) Williams, R. M. *Synthesis of Optically Active α -Amino Acids*; Pergamon Press: Oxford, 1989. (b) Williams, R. M.; Hendrix, J. A. *Chem. Rev.* **1992**, 92, 889–917. (c) Duthaler, R. O. *Tetrahedron* **1994**, 50, 1539–1650. (d) Cole, D. C. *Tetrahedron* **1994**, 50, 9517–9582.
3. See, for example: Hatanaka, S. I. *Progr. Chem. Org. Nat. Prod.* **1992**, 59, 1–141.
4. (a) Pandey, R. C.; Meng, H.; Cook Jr., J. C.; Rinehart Jr., K. L. *J. Am. Chem. Soc.* **1977**, 99, 5203–5205. (b) Pandey, R. C.; Cook Jr., J. C.; Rinehart Jr., K. L. *J. Am. Chem. Soc.* **1977**, 99, 5205–5206. (c) Bodo, B.; Rebuffat, S.; El-Hajji, M.; Davoust, D. *J. Am. Chem. Soc.* **1985**, 107, 6011–6017.
5. Gante, J. *Angew. Chem. Int. Ed. Engl.* **1994**, 33, 1699–1720.
6. For a recent, short review on the topic of α,α -disubstituted α -amino acids, see: Wirth, T. *Angew. Chem. Int. Ed. Engl.* **1997**, 36, 225–227.
7. Yokum, T. S.; Gauthier, T. J.; Hammer, R. P.; McLaughlin, M. L. *J. Am. Chem. Soc.* **1997**, 119, 1167–1168.
8. For the synthesis of peptides containing such types of amino acids, see: (a) Wipf, P.; Heimgartner, H. *Helv. Chim. Acta* **1990**, 73, 13–24. (b) Mickos, H.; Sundberg, K.; Lüning, B. *Acta Chem. Scand.* **1992**, 46, 989–993. (c) Toniolo, C.; Formaggio, F.; Crisma, M.; Valle, G.; Boesten, W. H. J.; Schoemaker, H. E.; Kamphuis, J.; Temussi, P. A.; Becker, E. L.; Précigoux, G. *Tetrahedron* **1993**, 49, 3641–3653. (d) Burger, K.; Hollweck, W. *Synlett* **1994**, 751–753. (e) Kaptein, B.; Monaco, V.; Broxterman, Q. B.; Schoemaker, H. E.; Kamphuis, J. *Rec. Trav. Chim. Pays-Bas* **1995**, 114, 231–238. (f) Bucher, C. B.; Linden, A.; Heimgartner, H. *Helv. Chim. Acta* **1995**, 78, 935–946. (g) Obrecht, D.; Lehmann, C.; Ruffieux, R.; Schönholzer, P.; Müller, K. *Helv. Chim. Acta* **1995**, 78, 1567–1587. (h) Palomo, C.; Aizpurua, J. M.; Ganboa, I.; Odriozola, B.; Urchegi, R.; Görls, H. *Chem. Commun.* **1996**, 1269–1270. (i) Sekizaki, H.; Itoh, K.; Toyota, E.; Tanizawa,

- K. *Tetrahedron Lett.* **1997**, 38, 1777–1780. (j) Scott, W. L.; Zhou, C.; Fang, Z.; O'Donnell, M. J. *Tetrahedron Lett.* **1997**, 38, 3695–3698.
9. (a) Schöllkopf, U.; Groth, U.; Hartwig, W. *Liebigs Ann. Chem.* **1981**, 2407–2418. (b) Seebach, D.; Aebi, J. D.; Gander-Coquoz, M.; Naef, R. *Helv. Chim. Acta* **1987**, 70, 1194–1216. (c) Williams, R. M.; Im, M.-N. *J. Am. Chem. Soc.* **1991**, 113, 9276–9286. (d) Altmann, E.; Nebel, K.; Mutter, M. *Helv. Chim. Acta* **1991**, 74, 800–806. (e) Kawabata, T.; Wirth, T.; Yahiro, K.; Suzuki, H.; Fuji, K. *J. Am. Chem. Soc.* **1994**, 116, 10809–10810. (f) Berkowitz, D. B.; Smith, M. K. *J. Org. Chem.* **1995**, 60, 1233–1238. (g) Studer, A.; Seebach, D. *Liebigs Ann. Chem.* **1995**, 217–222. (h) Ferey, V.; Toupet, L.; Le Gall, T.; Mioskowski, C. *Angew. Chem. Int. Ed. Engl.* **1996**, 35, 430–432. (i) Ma, D.; Tian, H. *Tetrahedron: Asymmetry* **1996**, 7, 1567–1570. (j) Badorrey, R.; Cativiela, C.; Díaz de Villegas, M. D.; Gálvez, J. A.; Lapeña, Y. *Tetrahedron: Asymmetry* **1997**, 8, 311–317. (k) Chinchilla, R.; Falvello, L. R.; Galindo, N.; Nájera, C. *Angew. Chem. Int. Ed. Engl.* **1997**, 36, 995–997.
10. Grandel, R.; Kazmaier, U.; Nuber, B. *Liebigs Ann. Chem.* **1996**, 1143–1150, and references therein.
11. (a) Bravo, P.; Viani, F.; Zanda, M.; Soloshonok, V. *Gazz. Chim. Ital.* **1995**, 125, 149–150. (b) Jung, M. E.; D'Amico, D. C. *J. Am. Chem. Soc.* **1995**, 117, 7379–7388. (c) Imogaï, H.; Petit, Y.; Larchevêque, M. *Tetrahedron Lett.* **1996**, 37, 2573–2576. (d) Westermann, B.; Gedrath, I. *Synlett* **1996**, 665–666. (e) Kazmaier, U.; Schneider, C. *Synlett* **1996**, 975–977.
12. (a) Colson, P. J.; Hegedus, L. S. *J. Org. Chem.* **1993**, 58, 5918–5924. (b) Hegedus, L. S. *Acc. Chem. Res.* **1995**, 28, 299–305. (c) Ojima, I. *Acc. Chem. Res.* **1995**, 28, 383–389.
13. Liu, W.; Ray, P.; Benezra, S. A. *J. Chem. Soc., Perkin Trans. 1* **1995**, 553–559.
14. (a) Moon, S.-H.; Ohfune, Y. *J. Am. Chem. Soc.* **1994**, 116, 7405–7406. (b) Cativiela, C.; Díaz de Villegas, M. D.; Gálvez, J. A.; García, J. I. *Tetrahedron* **1996**, 52, 9563–9574. (c) Kunz, H. In *Houben-Weyl's Methods of Organic Chemistry, Stereoselective Synthesis*; Helmchen, G., Hoffmann, R. W., Mulzer, J., Schaumann E., Eds.; Georg Thieme Verlag: Stuttgart, 1996; Vol. 3, pp. 1931–1952. (d) Horikawa, M.; Nakajima, T.; Ohfune, Y. *Synlett* **1997**, 253–254.
15. For stereoselective additions to C=N bonds, see pertinent references in the preceding article.^{16b} For the synthesis of α,α -disubstituted α -amino acids based on this methodology, see Ref. 11a and: (a) Harwood, L. M.; Vines, K. J.; Drew, M. G. B. *Synlett* **1996**, 1051–1053. (b) Moody, C. J.; Lightfoot, A. P.; Gallagher, P. T. *Synlett* **1997**, 659–660.
16. (a) Murga, J., Ph.D. Thesis. Jaume I, 1996. For a preliminary account, see: Marco, J. A.; Carda, M.; Murga, J.; González, F.; Falomir, E. *Tetrahedron Lett.* **1997**, 38, 1841–1844. The melting points reported in this paper for 2-methyl serine and 2-phenylserine were erroneously interchanged. (b) Marco, J. A.; Carda, M.; Murga, J.; Rodríguez, S.; Falomir, E.; Oliva, M. *Tetrahedron: Asymmetry* **1998**, 9, 1679–1701.
17. The additions of *t*-butyllithium and allyllithium to oxime (*E*)-**2** with R¹, R²=Bn, R³=TPS took place with a similarly high d.r.,^{16b} but the conversion of the obtained products into α -*t*-butyl serine and α -allyl serine has not yet been accomplished.
18. Bal, B. S.; Childers Jr., W. E.; Pinnick, H. W. *Tetrahedron* **1981**, 37, 2091–2096.
19. Williams, D. R.; Sit, S.-Y. *J. Am. Chem. Soc.* **1984**, 106, 2949–2954. Cleavage of acetonides (–)-**14** and (–)-**21** under the aqueous acidic conditions used for (–)-**7** provided not only the desired diols but also products of internal transacylation (oxazolidinone migration).
20. Henze, H. R.; Craig, W. C. *J. Org. Chem.* **1945**, 10, 2–9.
21. Ito, Y.; Sawamura, M.; Shirakawa, E.; Hayashizaki, K.; Hayashi, T. *Tetrahedron Lett.* **1988**, 28, 235–238.
22. Marco, J. A.; Carda, M.; González, F.; Rodríguez, S.; Murga, J. *Liebigs Ann. Chem.* **1996**, 1801–1810.
23. For recent references: (a) Hattori, K.; Yamamoto, Y. *Tetrahedron* **1994**, 50, 2785–2792. (b) Palomo, C.; Aizpurua, J. M.; Cabré, F.; García, J. M.; Odriozola, J. M. *Tetrahedron Lett.* **1994**, 35, 2721–2724. (c) Bunnage, M. E.; Davies, S. G.; Goodwin, C. J.; Ichihara, O. *Tetrahedron* **1994**, 50, 3975–3986. (d) Schmeck, C.; Hegedus, L. S. *J. Am. Chem. Soc.* **1994**, 116, 9927–9934. (e) Banfi, L.; Cascio, G.; Guanti, G.; Manghisi, E.; Narisano, E.; Riva, R. *Tetrahedron* **1994**, 50, 11967–11982.
24. For recent references: (a) Palomo, C.; Aizpurua, J. M.; Cabré, F.; Cuevas, C.; Munt, S.; Odriozola, J. M. *Tetrahedron Lett.* **1994**, 35, 2725–2728. (b) Cardillo, G.; de Simone, A.; Gentilucci, L.; Sabatino, P.; Tomasini, C. *Tetrahedron Lett.* **1994**, 35, 5051–5054. (c) Reetz, M. T.; Röhrig, D.; Harms, K.; Frenking, G. *Tetrahedron Lett.* **1994**, 35, 8765–8768.
25. Kolasa, T.; Sharma, S. K.; Miller, M. J. *Tetrahedron* **1988**, 44, 5431–5440.
26. For a recent reference, see: Greven, R.; Jütten, P.; Scharf, H.-D. *J. Org. Chem.* **1993**, 58, 3742–3747.